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values for aqueous potassium chloride solutions with molalities of 001 mol/kg, 01 mol/kg and 10 mol/kg at temperatures from 0°C to 50°C The values, which are based on measurements at the CRC Handbook of Chemistry and Physics, 91th Edition Author: Dr David Lide

CRC HANDBOOK OF LABORATORY SAFETY

CRC HANDBOOK OF LABORATORY SAFETY 4624 Perchloric Acid* 4624 1 Introduction Considerable interest has been taken in the explosion hazards to be encountered in the use of perchloric acid since a mixture of perchloric acid and acetic anhydride exploded in a Los Angeles factory in 1947, killing 15, injuring 400, and causing \$2 million in

Table of Specific Heat Values - Chemistry-Reference

Table of Specific Heat Values Substance State of Matter Cp (J/g°C) Aluminum Solid 0897 Benzene Liquid 174 Brass Solid 0380 Bromine Liquid 0226 Copper Solid 0385 Ethanol Liquid 244 Gold Solid 0129 Iodine Solid 0145 Iron Solid 0449 Isopropyl (Rubbing) Alcohol Liquid 261 Lead Solid 0129 Manganese Solid 0479 Methanol Liquid 253 Nickel

Physical constants of inorganic compounds

Physical constants of inorganic compounds The compounds in this table were selected on the basis of their laboratory and industrial importance, as well as their value in illus-

Handbook of Surface and Colloid Chemistry Fourth Edition

202 Handbook of Surface and Colloid Chemistry Thus, the regular solution theory is strictly valid for nonpolar slightly polar systems, without any specific interactions According to Equations 33 and 34, if solvent and polymer have the same solubility parameters, the heat of mixing is zero and they are therefore miscible at all proportions

Appendix A: Table of Densities of Water

Appendix A: Table of Densities of Water Density of Water at 1 atmosphere (from: Handbook of Chemistry and Physics, CRC press, 64th Ed)

Temperature °C Density g/mL Temperature °C Density g/mL

THE UNIVERSITY OF THE STATE OF NEW YORK ... - AP Chemistry

Reference Tables for Physical Setting/Chemistry - 2011 Edition 1 Reference Tables for Physical Setting/Chemistry - 2011 Edition 5 Table I Heats of Reaction at 1013 kPa and 298 K Table J Source: CRC Handbook of Chemistry and Physics, 91st ed, 2010-2011, CRC Press

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aqueous solubility of inorganic compounds at Various ...

aqueous solubility of inorganic compounds at Various temperatures The solubility of over 300 common inorganic compounds in water is tabulated here as a function of temperature Solubility is defined as the concentration of the compound in a solution that is in equilibrium with ...

PHYSICAL CONSTANTS OF INORGANIC COMPOUNDS

PHYSICAL CONSTANTS OF INORGANIC COMPOUNDS The compounds in this table were selected on the basis of their laboratory and industrial importance, as well as their value in illustrating trends in the variation of physical properties with position in the periodic table An effort has been made to include the most frequently encountered inor-

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Density of Water (g/mL) vs. Temperature (°C)

Density of Water (g/mL) vs Temperature (°C) (from Handbook of Chemistry and Physics , 53rd Edition, p F4) 00 01 02 03 04 05 06 07 08 09

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