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When alkali is added to beryllium salt solutions the β -form (a gel) is formed. If this left to stand or boiled, the rhombic α -form precipitates. [7] This has the same

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structure as zinc hydroxide , $Zn(OH)_2$, with tetrahedral beryllium centers.

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Two possible interactions can occur between the metal atoms on the electrode and the ion solutions. Metal ion M^{n+} from the solution may collide with the electrode, gaining "n" electrons from it, and convert to metal atoms. This means that the ions are reduced. Metal atom on the surface may lose "n" electrons to the electrode and enter the solution as the ion M^{n+} meaning that the metal ...

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Buffer solutions are used as a means of keeping pH at a nearly constant value in a wide variety of chemical applications. In nature, there are many systems that use buffering for pH regulation. Bulk modulus The bulk modulus (or) of a substance is a measure of how resistant to compression that substance is. It is defined as the ratio of

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the infinitesimal pressure increase to the resulting ...

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